1 Molecular basis of unidirectional information transmission in two-component

2 systems: lessons from the DesK-DesR thermosensor

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16 Abstract

17 Cellular signaling systems transmit information over long distances using allosteric 18 transitions and/or post-translational modifications. In two-component systems the 19 sensor histidine kinase and response regulator are wired through phosphoryl-transfer 20 reactions, using either a uni- or bi-directional transmission mode, allowing to build rich 21 regulatory networks. Using the thermosensor DesK-DesR two-component system from 22 *Bacillus subtilis* and combining crystal structures, QM/MM calculations and integrative

23 kinetic modeling, we uncover that: i) longer or shorter distances between the 24 phosphoryl-acceptor and -donor residues can shift the phosphoryl-transfer equilibrium; 25 ii) the phosphorylation-dependent dimerization of the regulator acts as a sequestering 26 mechanism by preventing the interaction with the histidine kinase; and iii) the kinase's 27 intrinsic conformational equilibrium makes the phosphotransferase state unlikely in the 28 absence of histidine phosphorylation, minimizing backwards transmission. These 29 mechanisms allow the system to control the direction of signal transmission in a very 30 efficient way, showcasing the key role that structure-encoded allostery plays in 31 signaling proteins to store and transmit information.

32 Introduction

33 Perception is a fundamental process that allows living organisms to adapt to highly 34 variable environments. Cells are equipped with a set of signaling pathways that 35 recognize specific signals, transmit and process the information to regulate different 36 cellular programs. Two-component systems (TCS) are relevant constituents of this 37 sensorial system in bacteria and archaea, also found in fungi and plants¹. TCSs are 38 usually composed of a sensor histidine kinase (HK), which detects a specific signal that 39 allosterically regulates their catalytic activities. Information is then transmitted further 40 downstream via phosphorylation of a specific response regulator (RR)², which acts as 41 the second component of the TCS pathway, executing the output response.

42 Upon the signal-dependent pathway activation, the HK autophosphorylates a 43 conserved His. A second reaction takes place in tandem, and the phosphoryl moiety 44 gets transferred to an invariant Asp within its cognate RR. Activation of the RR by 45 phosphorylaton usually promotes its dimerization and subsequent binding to DNA, as 46 most RRs bear a DNA-binding domain responsible for exerting the effector response by 47 changing the expression of target genes³. In most TCSs, when the signal is absent, the 48 HK also promotes the dephosphorylation of the RR (*i.e.* acting as a phoshatase), 49 making of most HKs fascinating examples of paradoxical enzymes⁴. HK-mediated 50 P~RR dephosphorylation has been shown to be an extremely relevant process for 51 shutting down the system⁵ and preventing crosstalk⁶ in vivo. Interestingly, TCS 52 pathways have also evolved to increase their complexity by wiring additional 53 intermediate RR and His containing phosphotransferase proteins (HPt) domains that 54 ultimately build phosphorylation cascades, known as phosphorelays.

55 In the past few years several groups have made important contributions to the understanding of the regulation of the activities of HK and RR^{3,7-14}, how their 56 components are functionally wired through specific protein-protein interactions¹⁵ and 57 58 how the signal is transmitted from the extra- to the intra-cellular space^{16,17}. Presently, 59 five HK families (HisKA, HisKA 3, HisKA 2, H-kinase dim and HWE HK) have been 60 recognized based on sequence clustering¹⁸; these families share many similarities at the 61 structural level^{9,11,13,19-21}. However, we are still far from understanding several 62 fundamental aspects of how HK-driven phosphorylation cascades transmit and process 63 information efficiently. For example, are phosphorylation/dephosphorylation futile 64 cycles minimized to avoid cell energy dissipation? If so, how is this accomplished from 65 a structural and mechanistic perspective? Or yet, how is the directionality of 66 phosphoryl-transfer reactions enforced, such that the P~RR species are accumulated, 67 or even to allow the connection with additional intermediate components to build richer 68 phosphorelays pathways?

69 Phosphoryl-transfer reactions occur by nucleophilic substitution. The electron-rich

oxygen of the RR's Asp carboxylate, attacks the phosphorus electrophile, substituting the ε -nitrogen of the HK's His imidazole²² (Extended Data Fig 1). The directionality issue should not be overlooked, since His phosphorylation is expected to be favored over Asp phosphorylation considering available data on the hydrolysis of phosphoramidate and phosphoanhydride compounds²³⁻²⁵. Dedicated mechanisms are thus anticipated to surmount the uphill direction and ensure proper physiologic behavior.

77 The TCS DesK-DesR from Bacillus subtilis is a thermosensor pathway that provides the 78 cells with a fast homeostatic response that maintains cell membrane fluidity²⁶. DesK is a 79 HisKA 3 HK that is activated upon cold shock. Phosphorylation at the invariant D54_{RR} 80 (Asp at position 54) of its cognate RR DesR triggers the adaptive response by activating 81 the expression of a fatty acid desaturase²⁷. The structural information available for this 82 system is unique. Several experimental structures of both components in different steps 83 of the signaling pathway, and adopting different conformations have been 84 elucidated^{7,9,28,29}. The catalytically active cytosolic and soluble region of DesK (DesKC) 85 can adopt at least three functionally and structurally different conformations: 86 autokinase, phosphatase phosphotransferase and states, involved in the 87 autophosphorylation, phosphoryl-transfer and dephosphorylation activities. 88 respectively. Based on the crystal structures of the DesK:DesR complex, and their 89 comparison to other TCSs, the distance between the phosphoryl-donor and -acceptor 90 residues (the HK His and RR's Asp, respectively) correlates with the systems' 91 directionality or degree of phosphoryl-transfer reversibility. Shorter distances predict a 92 tight transitions states in the nucleophilic substitution reaction²², *i.e.* a pentavalent 93 bipyramidal transition state, and for still not understood reasons correlates with more reversible phosphoryl-transfer reactions. Inversely, longer distances between reactive residues imply loose transitions states, *i.e.* a more dissociated planar trigonal metaphosphate transition state, that correlate with irreversible P~His \rightarrow Asp transfer reactions⁷.

98 In the present work, to further understand how phosphoryl-transfer reversibility is 99 controlled in TCSs, quantum mechanics/molecular mechanics (QM/MM) calculations 100 were performed describing the phosphoryl-transfer (phosphoryl transfer from His to Asp 101 or Asp to His) and dephosphorylation reactions (phosphoryl-hydrolysis or transfer 102 reaction from Asp to a water molecule) using the DesK-DesR system. The crystal 103 structure of the DesK:DesR complex in the phosphatase state was refined to higher 104 resolution than that previously available, further supporting the QM/MM data. Critical 105 residues for the phosphotransferase and the phosphatase activities were pinpointed 106 utilizing structure-guided point mutants. Finally, an integrative structural and kinetic 107 model of the system was constructed, identifying key directionality determinants of the 108 phosphoryl-transfer reaction. Overall, evidence that the different HK enzymatic activities 109 are insulated from each other is provided, a pivotal element that ensures information is 110 efficiently transmitted in the right direction.

111 **Results**

112 A hydroxyl anion is required for phospho-aspartate hydrolysis in the HK-mediated

113 phosphatase reaction

114 To explore the mechanistic details of the dephosphorylation reaction in TCSs, the Free 115 Energy Profile of the DesK-DesR system was calculated using QM/MM-based steered 116 molecular dynamics. Calculations were based on an improved version of the crystal 117 structure of the DesK:DesR complex in the phosphatase state (PDB id: 7SSJ), used as 118 the starting species. The refinement of the previously reported structure was improved by reprocessing the raw diffraction data³⁰ with now available algorithms that better 119 handle anisotropic diffraction³¹. The resolution could thus be extended to 2.52 Å in the 120 121 best direction, and 2.8 Å in the other two (Extended Data Table S1). The re-refined 122 electron density maps improved substantially. Among other features, Q193_{HK} (GIn at 123 position 193 of the HK) was now well defined in density, interacting with a water molecule that coordinates the Mg²⁺ cation (Fig. 1a). An additional water molecule is now 124 125 clearly visible, hydrogen-bonded to the conserved T80_{BB} side chain (Fig. 1a), and well 126 positioned to interact with the attacking water molecule for phosphoryl hydrolysis. 127 Indeed, an unexpected electron density bulge coinciding with a Fourier difference peak 128 was observed near the phospho-mimetic BeF3⁻ group, axially in line with the Be-OD54 129 bond to D54_{BB} (Extended Data Fig 2). This feature is consistent with a water molecule 130 correctly placed to perform the nucleophilic attack, although limited data resolution 131 precludes conclusive modeling.

132 Our results showed that in order to simulate the P~Asp_{BB} hydrolysis reaction, the 133 attacking nucleophile has to be a hydroxyl anion, and not a water molecule. Attempts to 134 perform the reaction using a water molecule as the phosphate acceptor were 135 unsuccessful, even if the transferring phosphate was concertedly probed as the proton 136 acceptor. It must be stressed that the reaction center lacks a suitable residue that may 137 act as a base to deprotonate the reactive water. Therefore, the hydroxyl must be 138 formed in the bulk solvent. This observation is in accordance with the well-studied 139 Ras/GAP proteins mediating GTP hydrolysis, where a solvent-assisted mechanism has 140 indeed been put forward³².

141 The HK-mediated dephosphorylation reaction proceeds through a concerted 142 nucleophilic substitution mechanism, as evidenced by the changes at the interactomic 143 distance in the TS (transition state) zone (Fig. 1b and c), with an energy barrier of ~23 144 kcal/mol (Fig. 1b and c). The bond with the attacking water molecule occurs late in the 145 TS (defined as the higher energy state of the reaction pathway), implying a dissociative 146 mechanism with loose TS. The TS adopts the expected planar trigonal structure. 147 $Q193_{HK}$, which is bound to the Mg coordination sphere, remains hydrogen bonded to, 148 and accompanying the acceptor OH⁻ along the reaction, only to release it after the TS is 149 resolved, and the orthophosphate liberated as final product. The same QM/MM 150 calculation was performed in the absence of Mg²⁺, displaying a much higher energy 151 barrier. On the other hand, the calculated reaction without DesK, surprisingly showed a 152 similar profile (Extended Data Fig 3), suggesting that additional mechanisms should be 153 at play. Taken all the evidence together, the sidechain amide of the conserved Q193 on 154 the HK's al helix, appears to assist in the reaction by placing the hydroxyl anion in a 155 catalytically ideal in-line attack position, which would otherwise be diverted by the 156 action of the phosphoryl oxygens³³.

157 **DesK:DesR phosphoryl-transfer follows a dissociative nucleophilic substitution**

158 mechanism

159 To test whether the difference in the distances between phosphoryl-acceptor and -160 donor residues correlates with a looser or tighter phosphoryl-transfer TS between 161 phosphohistidine and aspartate, the free energy profiles of the reactions were 162 computed using QM/MM-based steered molecular dynamics using either a HK:RR or a 163 phosphorelay Hpt:RR type of complexes. First, build to the starting 164 phosphotransferase-competent state of the DesK:DesR complex, the crystal structure 165 of the $DesKC_{H188E}$: $DesR_{REC}$ ¹ complex was used (PDB id:5IUK)⁷, into which the 166 phosphorylated H188_{HK} was modeled by superimposing the crystal structure of 167 phosphorylated wild type DesK alone (PDB id:5IUM)⁷. The reaction could be nicely 168 simulated, exhibiting a low energy barrier (Fig. 2a, Table 1 and Extended Data Fig 4), 169 and confirming that the starting point is a good representation of the 170 phosphotransferase state.

171 The reaction starts with the phosphoryl molety bound onto $H188_{HK}$ and also 172 establishing a hydrogen-bond with Thr 80_{BB} (Extended Data Fig 4b). The latter hydrogen 173 bond remained during the whole reaction. The TS shows the en transfer phosphoryl-174 group adopting a planar trigonal structure (metaphosphate). Remarkably, the TS zone 175 shows a dissociative character. Although the reactive D54_{BB} got initially closer to the 176 phosphoryl group (Fig. 2a), the O-P distance stopped decreasing at ca. 2.8 Å (i.e. the 177 new covalent bond is not yet established) to then remains constant while the N-P bond is stretched. Only after the donor N-P bond increased to 3.1Å (*i.e.* bond cleavage) the 178 179 D54_{BB} O-P bond was properly formed (Fig. 2a and Extended Data Fig 4b).

180 The role of the Mg²⁺ was assessed by calculating the free energy profile in the absence 181 of the metal cation. This resulted in a much larger energy barrier and unfavorably 182 positive reaction free energy (ΔG°), similar to the calculated phosphoryl-transfer 183 reaction between isolated His and Asp residues in water (Table 1 and Extended Data Fig 4a). The positive ΔG^0 for the isolated reaction is expected from a pure chemical 184 185 viewpoint, as described in the introduction. Consistently, phosphoryl-transfer reactions 186 using equimolar amounts of $DesR_{REC} \sim P$ and DesKC, in the absence of Mg^{2+} showed 187 that the reaction proceeds slowly toward the His (Fig. 2b), spontaneously approaching

¹ REC: the phosphorylatable receiver domain of response regulators

maximum HK phosphorylation (*i.e.* hemiphosphorylation of the HK dimer)²⁹. This indicates that the phosphoryl-transfer reaction itself is not completely abolished in the absence of Mg^{2+} , but the equilibrium is highly displaced toward the His.

191 Secondly, to compare bona fide HK:RR complexes with those of phosphorelay 192 pathways, similar free energy profile analyses were computed using the yeast 193 osmosensor phosphorelay complex Ypd1:Sln1 (Fig. 2). In this case the reaction showed 194 a more associative mechanism than for the DesK-DesR system, consistent with an 195 initial shorter distance between the phosphoryl donor and acceptor residues, and a 196 similarly small energy barrier of 7 kcal/mol (Table 1). The difference between the 197 product free energy (phosphorylated Asp) and the starting point (phosphorylated His) is 198 more positive in Ypd1:Sln1 as compared to the DesK-DesR system (Table 1 and 199 Extended Data Fig 4a) and similar to the His:Asp reaction in water (Table 1). This is an 200 important observation, consistent with our hypothesis of a correlation among 201 associative phosphoryl-transfer reactions, shorter inter-atomic distances, and the 202 chemical equilibrium displaced toward His phosphorylation, and vice versa. Altogether, 203 our results thus provide strong evidence that the experimental crystal structure of the 204 $DesK_{H188F}$: $DesR_{RFC}$ complex is a good mimetic of the phosphotransferase state and that 205 phosphoryl-transfer proceeds through a more dissociative mechanism or loose 206 transition state in DesK:DesR than in the phosphorelay.

Highly conserved D189_{HK} and T80_{RR} residues are not essential for phosphoryl transfer

The configuration of the phosphotransferase active site places the highly conserved $T80_{BB}$ at interaction distance with the covalently bound phosphoryl moiety of H188_{HK}.

211 This interaction was kept during the entire phosphoryl-transfer process in our QM/MM 212 calculations. To further analyze the relevance of T80_{RB} in the reaction, the HK-mediated 213 phosphorylation of T80A_{BB} (Fig. 3a) and T80S_{BB} (Extended Data Fig 5) mutants were 214 measured. In the *in vitro* assays, phosphoryl-transfer and desphosphorylation of the RR 215 are simultaneously taking place (biphasic curves). Intriguingly, DesK was able to 216 phosphorylate these two DesR mutants, showing phosphoryl-transfer reaction 217 comparable to wild type (Fig. 3b), thus indicating that T80_{RR} is not critical for the 218 reaction. QM/MM calculations of the phosphoryl-transfer reaction using the T80A 219 mutant, further suggest this residue is not essential (Table 1). Of note, T80A_{RB}, but not 220 $T80S_{BR}$, was unable to be phosphorylated using acetyl-phosphate as a phosphoryl-221 donor (data not shown).

222 Concerning D189_{HK}, it can be predicted that this highly conserved acidic residue at 223 position H+1 (one residue C-terminal to the phosphorylatable His), is not involved in the 224 phosphoryl-transfer reaction. This is in contrast with the critical role that this residue 225 plays in subtracting a proton from the His δN , increasing the nucleophilicity of the His in 226 the autophosphorylation reaction^{10,13,34,35}. In the phosphotransferase state, due to the 227 rotameric configuration of H188_{HK}, the δN is oriented in a way that cannot interact with 228 D189_{HK}. Consistent with this view, reverse phosphoryl-transfer of a D189A_{HK} mutant 229 was not abolished (Fig. 3c).

Phosphoryl-transfer reversibility is modulated by amino acid substitutions on theRR

232 QM/MM calculations suggested that, although a looser TS could be associated with a 233 shift in the equilibrium towards phosphorylation of the RR, when compared to the more

associative mechanism of the phosphorelay system, the ΔG^{0} of the reaction was still 234 235 positive. Thus, on thermodynamic grounds, the transfer reaction appears to always 236 favor the phosphorylation of the HK. In this scenario, the following question emerges: 237 Have TCSs evolved additional mechanisms to deal with this energetic uphill? Three 238 different mechanism could be at play: i) dimerization of the phosphorylated form of 239 DesR could shift the equilibrium towards its phosphorylated state, especially 240 considering that its a1a5 surface is used for both RR dimerization and DesK-241 interaction; ii) phosphorylated DesR might exhibit decreased affinity for DesK when 242 compared to the unphosphorylated species; and/or iii) in the absence of the sensor 243 domain, a shifted conformational equilibrium of DesK towards the phosphatase-244 competent state would reduce the amount of available phosphorylatable His, given that 245 it is occluded inside the **D**imerization and **H**istidine **p**hosphotransfer domain (DHp)⁷. 246 Since the interaction surfaces between DesK and DesR, comparing the HK's 247 phosphatase- and phosphotransferase-competent states, are very similar, as are the 248 conformations of bound DesR, the second option of affinity changes seems unlikely. 249 Also, if DesR dimerization were to sequester the phosphorylated monomeric DesR 250 species out of equilibrium, this should also preclude DesK-mediated 251 dephosphorylation, which is clearly not the case. Thus, although dimerization could 252 contribute to the observed P~His_{HK} \rightarrow Asp_{BR} directionality, it seems not be the main 253 driving force.

We reasoned that if the phosphatase/kinase equilibrium of DesKC is shifted towards the former state, mutating residues that selectively affect DesR's interaction with the phosphatase-competent state of the kinase only would induce a shift in phosphoryltransfer reversibility. To test this hypothesis two DesR mutants, R84A_{RB} and Q10A_{RB}, were tested. Residue R84_{RR} interacts specifically with DesK's D189_{HK}⁷, and due to the gear-box mechanism⁹, the R84_{RR}:D189_{HK} interaction is disrupted in the phosphotransferase state through the rotation of the DHp domain α -helices (Fig. 4a). The R84A_{RR} substitution significantly reduced the dephosphorylation of the P~DesR compared to wild type (Fig 3d), but the phosphoryl group still accumulates in DesR (Figure 4b).

264 On the other hand, residue $Q10_{BB}$ is inserted in a pocket generated at the interface 265 between the DHp and ATP binding domain (Figure 4c), a hallmark of the phosphatase 266 state³⁶. Remarkably, a Q10A_{BB} mutant showed a significant increase in phosphoryl-267 transfer reversibility, with the phosphoryl molety more evenly distributed between the 268 kinase and the regulator (Fig. 4d). $Q10_{RR}$ is far from the phosphorylation site and was 269 previously highlighted as highly covariant with Q193 on the HK^7 . To rule out that 270 transfer reversibility of the Q10A_{BB} mutant is caused by a different positioning of DesR 271 in the phosphotransferase complex, potentially shortening the His_{HK}-Asp_{RR} distance, the 272 crystal structure of this complex was solved at 3.4 Å resolution. The structure clearly 273 indicated that the REC domain of DesR remains in the same position compared to the 274 wt (Extended Data Fig 6 and Table S1), especially not altering the His-Asp distance. The 275 observed change in phosphoryl-transfer reversibility is thus based on other reasons, 276 and directionality can be shifted by mutations at the response regulator that are far 277 from the phosphorylation site.

278 A shifted HK conformational equilibrium minimizes phosphoryl-transfer reversal

To better understand the key determinants of information transmission, a systemsbiology and integrative approach was followed. Taking advantage that the different

281 functional states of DesK can be trapped³⁷, an experimental dataset comprising 770 282 measurements of the level of phosphorylation level in each protein (HK or RR) was 283 generated at different time points along the phosphoryl-transfer reaction assays. Two 284 types of reactions were analyzed, either starting with the phosphorylation of the HK 285 (Fig. 5c,d,e,f and g) or the RR (Fig. 5h and Extended Data Fig 7a,b,c,d, e and g). To this 286 end we used DesKCwt, which lacks the transmembrane sensor domain, and was 287 previously shown to be deregulated^{9,38}. DesKC_{0193A} was used to evaluate phosphoryl-288 transfer with no confounding phosphatase activity. Q193_{HK} (H+5 position) is well 289 conserved in all HisKA_3 HKs and critical to keep phosphatase activity^{39,40}. DesKC_{DEST} 290 was also included, since this is a mutant that shifts the conformational equilibrium of 291 DesK towards the auto-kinase active state, by destabilizing the N-terminal coiled-coil⁸. 292 DesKC_{H188E} and DesKC_{H188V} are constructs that trap DesK in the phosphotransferase^{7,9} 293 and phosphatase state³⁸, respectively. In addition, wild type DesR_{BEC} and the point 294 mutant DesR_{REC-010A}, were also tested. The analyzed reactions confirmed several 295 experimentally observed behaviors, such as the decrease in phosphatase activity of 296 DesKC_{0193A} (Fig. 5e and Extended Data Fig 7c), or the increased phosphoryl-transfer 297 reversibility of DesR_{BEC-010A} (Fig. 5d and f and Extended Data Fig 7b and d).

Different kinetic models were constructed and optimized, taking into consideration the ensemble of available structural and biochemical data. The best-fitted model includes 19 free parameters (depicted in Fig 5a) and converged to a unique solution (Fig 5b). Residuals to test the significance of the differences among alternative models were rigorously monitored (see methods).

303 Several salient conclusions can be drawn from the optimized integrative model: i) the 304 equilibrium constant of the phosphotransferase reaction is directed towards His_{HK} 305 phosphorylation, as expected from our QM/MM calculations (K_{phos} (k4/k3) = 20.7, Table 306 1); ii) the kinetic rate constant ($k5 = 0.227 \text{ s}^{-1}$) of the HK-mediated phosphatase activity 307 is 3 orders of magnitude higher than the intrinsic dephosphorylation rate of the RR (k9 =308 3.87E-4 s⁻¹, autodephosphorylation); iii) the Q193A_{HK} substitution reduces the 309 phosphatase activity by 2 orders of magnitude ($k5q = 6.082E-3 s^{-1}$), but the activity is 310 not completely abolished: iv) the dissociation equilibrium constant of P~RR dimerization 311 is in the low μ M range, and the Q10A_{BB} amino acid replacement increases this K_D by a 312 factor of 4 ($K_{D RR Q10A} = 1.442 \mu M$); v) the phosphatase/auto-kinase conformational 313 equilibrium of the HK is only slightly displaced towards the auto-kinase state; vi) 314 Q193A_{HK} and DesKC_{DEST} variant shift the phosphatase/auto-kinase equilibrium toward 315 the auto-kinase state; and, interestingly, vii) the auto-kinase/phosphotransferase 316 equilibrium is highly shifted towards the auto-kinase state, preventing phosphoryl back-317 transfer.

318 The kinetic model also predicts that DesKC will interact with DesR mostly in a 319 phosphatase conformation. Isothermal titration calorimetry (ITC) data indeed confirmed 320 this hypothesis. DesKC associated to DesR with a 2:2 stoichiometry (two RR molecules 321 per HK dimer) as had been previously observed using a phosphatase-trapped mutant⁷. 322 Moreover, the relatively similar K_D values of the two independent binding sites was 323 consistent with a symmetric architecture as evidenced in the phosphatase configuration 324 (Extended Data Fig 8). The phosphotransferase configuration, due to its asymmetry, 325 exhibits substantially larger differences in the K_{D} values⁷. The kinetic model allows us to 326 make further predictions, such as a shift of the dimerization equilibrium in the DesR 327 Q10A_{BB} mutant, driving it towards the monomeric species (higher dissociation constant 328 of dimerization K_{DRB010} compared to wild type). Size-exclusion chromatography of phosphorylated DesR-REC_{Q10A} indeed showed disturbed dimerization for this mutant compared to the wild type protein (Extended Data Fig 9). Taking all the evidence from the model together, we conclude that phosphoryl-transfer directionality is dictated by: i) the auto-kinase/phosphotransferase conformational equilibrium, ii) the ratio between the forward and reverse phosphoryl-transfer kinetic rates, and iii) the dimerization of the $P\sim RR$.

335 **Discussion**

336 A subtle displacement of the RR in the HK:RR interaction minimizes futile cycles

337 Signal transduction pathways that use phosphorylation as a means to transmit 338 information use cellular energy by consuming ATP. In this context a key question is: 339 Have molecular machines evolved to minimize unwanted energy waste? There are 340 examples where energy dissipation is biologically relevant, for instance to deal with 341 noisy signaling and allow for adaptation⁴¹. In the case of TCSs, HK bifunctionality 342 confers robustness to the signaling process^{42,43}, but raises the guestion of how tightly it 343 must be regulated to avoid futile cycles and energy waste. In the present work we 344 described from first principles the HK-mediated dephosphorylation reaction of P~RR. 345 The crystal structure of the DesK:DesR phosphatase complex shows well-defined 346 density of key residues. That the phospho-mimetic BeF_{3} group bound to $D54_{BB}$ shows 347 an unexpected bulge in the electron density suggest that instead a MgF₃⁻ (magnesium 348 fluoride) could be present as mimetic of the transition state ⁴⁴ (Extended Data Fig 2). In 349 any case, the architecture of the reaction center in the X-ray structure is perfectly 350 consistent with the QM/MM calculations. The HK favors hydrolysis of the P~Asp by 351 organizing the active site within the RR, assisting with its glutamine in position H+5 in 352 the correct placement of the attacking hydroxyl anion. The interaction between the HK 353 in the phosphatase conformation and the phosphorylated RR promotes the opening of the RR active site, exposing the phosphorylation site^{7,28}. According to our kinetic model, 354 355 the dephosphorylation of the RR is accelerated by the HK by 3 orders of magnitude 356 compared to the intrinsic autodephosphorylation reaction. Moreover, in the absence of 357 the conserved Q193_{HK} (Q193A substitution), the reaction is 2 orders of magnitude 358 slower compared to wild type HK. This modest acceleration and the effect of Q193 359 mutation is consistent with the proposed role of Q193_{HK}. The QM/MM calculations of 360 the dephosphorylation reaction in the presence or absence of the HK are similar, 361 implying that the acceleration promoted by the HK is likely gained through lowering the 362 entropy of the Michaelis complex. Analogous mechanisms have been put forward after 363 extensive studies of the Ras/Ras-like small GTPase family of proteins. In the case of 364 Ras, a GTPase activating protein (GAP) accelerates the reaction by several orders of 365 magnitude⁴⁵. The active site is very similar to the reaction center conformed between 366 the HK and the RR in TCSs (Extended Data Fig 10). GAP binding to Ras induces a 367 significant loss of entropy of the GIn, which was proposed to be critical in the GTP 368 hydrolysis reaction mechanism³².

When the HK is stabilized in its phosphotransferase conformation, to avoid futile cycles the dephosphorylation of P~RR must be minimized. From a structural point of view, even though differences in the HK active site are observed comparing its phosphatase and phosphotransferase conformations, the position of Q193_{HK} is unchanged and the conformation of the RR is almost identical⁷. Then, why is there no significant phosphatase activity? An explanation can be uncovered by superimposing the DesK:DesR phosphatase and phosphotransferase complexes. There is a small but 376 significant shift (approximately 1 Å) in the position of the RR. This movement, in the 377 phosphatase state, brings the phosphoryl-group and the Mg²⁺, with its entire 378 coordination sphere, closer to Q193_{HK} (Extended Data Fig. 11). Hence, Q193_{HK} comes 379 now within interaction distance with a water molecule of the Mg²⁺ coordination sphere, 380 properly placed to position the attacking hydroxyl anion. The ~1Å shift of the RR is 381 driven by a set of interactions between the REC domain and the HK ATP binding 382 domain, that are only available in the phosphatase state. In contrast, in the 383 phosphotransferase state, Q193_{HK} is not ordered, as evidenced by the electron density, 384 which is not well defined (Extended Data Fig 11). Collectively, our data suggest that 385 interactions involving secondary interfaces in the phosphatase state, more precisely 386 between the REC and ATP binding domain, guide the correct placement of the RR to 387 favor dephosphorylation. In the absence of this interaction, the RR is held farther apart 388 and thus not promoting the phosphatase reaction.

389 Insulation of HK activities as an information-driving mechanism

390 The higher energy of the mixed anhydride O-P bond on the P~Asp compared to the N-P phosphoramidate in P~His²⁵, which is in turn higher than the anhydride O-P bond of 391 392 ATP⁴⁶, imposes an uphill energy barrier that the system must overcome to promote RR 393 phosphorylation. A shift in the phosphoryl-transfer reaction equilibrium according to the 394 distance between nucleophile and leaving group amino acids could be possible, 395 considering the position of the Mg²⁺ cation. The relative position of the cation changes 396 with respect to the HK, in different complex geometries. Our QM/MM calculations of phosphoryl-transfers along looser or tighter TS, suggest that changing the distance 397 398 between the donor and acceptor residues can indeed setup different AG of the 399 reaction. Nevertheless, the ΔG is always positive in the P~HK \rightarrow RR direction, so that 400 additional mechanisms are at play.

401 The structural model of the DesK-DesR system (Fig. 6a) suggests that the proper 402 accumulation of P~RR could be reached by: i) P~RR dimerization; ii) a reduction in 403 binding affinity to the HK when the RR is phosphorylated; and iii) through regulation of 404 the conformational equilibrium of the HK, making the phosphotransferase conformation 405 unlikely. According to the integrative kinetic model, the binding affinity of the HK to the 406 RR in phosphorylated and non-phosphorylated states of the latter, and either in the 407 phosphatase or phosphotransferase state of the HK, is unchanged. This is consistent 408 with all the experimental information, in particular the similarity of the phosphatase and 409 phosphotransferse binding areas at the HK:RR interface. Moreover, a decrease in the 410 affinity to the phosphorylated RR would also imply an inconsistent scenario where the 411 HK in the phosphatase state would be unable to interact with it. P~RR dimerization 412 does compete with HK binding, and the observed effect of the Q10A_{RR} mutant shows 413 that disturbing this equilibrium indeed results in a more reversible phosphoryl-transfer. 414 However, too tight a dimerization would also preclude HK-dependent and independent 415 dephosphorylation, which are constraints that need to be considered.

416 From our kinetic model, the parameters that contribute the most to the directionality 417 are: the ratio between the kinetic rates of the phosphotransferase along its forward and 418 reverse directions, and the conformational equilibrium between the HK's auto-kinase 419 and phosphotransferase states (Extended Data Fig 12). By moving the reactive His 420 closer or farther apart in the reaction center, the ΔG might be modulated due to the key 421 contribution of the Mg²⁺ cation at the beginning of the reaction, stabilizing the phospho-422 His in case the distance is short enough (Fig. 6b). This is in agreement with the ΔG 423 calculated for the isolated His to Asp phosphoryl-transfer reaction in water (Table 1),

424 suggesting that in an associative mechanism the influence of the Mg²⁺ in shifting the 425 equilibrium towards Asp phosphorylation is lower. The solvent-occluded position of the 426 His inside the DHp core of the HK in the phosphatase conformation⁷ should minimize 427 reversibility as well. Mutants that shift the phosphatase/auto-kinase equilibrium should 428 be more reversible. We did not observe such effect when we tested $DesK_{DEST}$ (Fig. 5 429 and Extended Data Fig 7). On the other hand, our finding that the equilibrium between 430 the auto-kinase and phosphotransferase conformations was shifted against the latter, 431 which is the only competent state able to participate in the phosphoryl-transfer 432 reaction, also explains why the His is not accessible anymore (Fig. 6c). From a 433 structural point of view it appears that phosphorylation favors the phosphotransferase 434 state by triggering a conformational rearrangement. The highly conserved R235 and 435 K242 in HK helix a2 are well positioned to interact with the phosphoryl-group on the 436 reactive H188. This interaction likely pulls the a2-helix toward the a1-helix, breaking the 437 symmetric organization of the DHp and favoring the phosphotransferase state (Fig. 6d). 438 A tight non-covalent interaction between the guanidinium group of the Arg with the 439 phosphate⁴⁷ could provide the necessary energy to promote this somehow unfavorable 440 conformational transition.

The unexpected conformational equilibrium uncovered by our kinetic model also indicates that the system is tuned to maximize signal transmission efficiency. In other words, in the presence of the specific signal, HK activation promotes a shift in the phosphatase/kinase conformational equilibrium towards the latter. However, depending on the kinase/phosphotransferase conformational equilibrium and intracellular protein concentrations, HK autophosphorylation might be inhibited by its the interaction with the RR in the phosphotransferase state. Our kinetic model predicts that there is a 448 narrow window of the kinase/phosphotransferase equilibrium in which the system can

449 accumulate phosphorylated RR, and requires that in the absence of phosphorylation,

450 the phosphotransferase state of the HK is inaccessible (Extended Data Fig 13).

451 Taking all together, we conclude that the sensor domain controls the on/off switch by 452 changing auto-kinase/phosphatase transition. But, the the auto-453 kinase/phosphotransferase equilibrium guarantees that the information cannot go 454 backwards. Finally, our findings might also be extrapolated to phosphorelay pathways, 455 in which different mechanisms could allow these systems to be more or less reversible. 456 by tuning the Asp-His distance, the dimerization of the phosphorylated effector RR, or 457 additional elements like local physicochemical properties in the immediate surroundings 458 of the P~Asp that were not considered in this work.

459 **Methods**

460 Classical (MM) Simulations

461 All molecular dynamics simulations were performed using the AMBER suite package of 462 software⁴⁸. Classical force field parameters for aminoacids were obtained from the 463 ff14SB force field, whereas force field parameters for phosporylated residues (phospo-464 Asp and phospho-his) were generated with the Antechamber module of the AMBER 465 suite, after geometry optimization and RESP charge derivation at the HF/6-31G* level 466 with Gaussian g09, revision D.01⁴⁹. For each simulation, the equilibration protocol consisted of 200-cycle runs of minimization with a 100 (kcal/mol)/Å² restraint constant 467 468 applied to the protein in order to relax the solvent structure, followed by a 1000-cycle 469 energy optimization in which the restraint was removed to avoid initial unfavorable 470 contacts. The system was then slowly heated to 300 K during a 1 ns simulation, with 471 the Berendsen thermostat. Finally, pressure was equilibrated at 1 atm over 1 ns, to let 472 the system reach the proper density. For all simulations we used the periodic boundary 473 condition approximation, with the Ewald summation method with a 10 Å cutoff for 474 nonbonded interactions, and the SHAKE algorithm for all hydrogen-containing bonds. 475 Final production 10 ns MD simulations were performed at 300 K using the Langevin 476 thermostat and a 2 fs time step, from which the last structure was selected for QM-MM 477 simulations.

478 Hybrid (QM-MM) Simulations

All DFT QM/MM calculations in this work were performed with the SANDER(AMBER) program and the QM(DFT)/MM implementation called LIO⁵⁰ using the PBE functional and a DZVP gaussian basis set. All relevant parts of the computation of LIO are ported to GPU, obtaining an improvement on the performance, including exchange and correlation, Coulomb, and QM/MM coupling terms^{51,52}.

For the phosphotransfer reactions, the QM region consisted of 70 atoms (Extended Data Fig 14), including both donor and acceptor amino acids (His/Asp), the magnesium ion and its whole coordination sphere (sidechain of $D9_{RR}$ and $D54_{RR}$, backbone carbonyl of $E56_{RR}$, two water molecules, and the phosphate) and a total of 8 H-link atoms at each corresponding boundary. For the phosphatase reaction the QM region is the same but with the donor His replaced by a hydroxyl anion.

490 Multiple Steered Molecular Dynamics protocol

491 To study both the phosphotransfer and the phosphatase reaction mechanisms, we 492 used a multiple steered molecular dynamics (MSMD) strategy, combined with 493 Jarzynski's relationship to determine the corresponding Free Energy Profiles. This 494 strategy has already been shown to be useful in previous works from our group for phosphoryl transfer reactions^{35,53,54}. Briefly, in MSMD the system is driven "multiple" 495 496 times along the selected reaction coordinate under nonequilibrium conditions, by 497 applying an external force, and for each individual trajectory, the work performed by the 498 external force is computed. Finally, multiple works are exponentially averaged using 499 Jarzynski's relationship to obtain the free energy profile. In the present work we first 500 performed 5 ns of conformational sampling at the reactive/product equilibrium 501 geometries using standard QM/MM Molecular Dynamics. This was followed by 10 502 independent MSMD simulations (starting from corresponding initial structures each 503 separated by 500 ps) which were run for approximately 2 ps using a 0.001 ps time step, 504 resulting in a pulling speed of 2 Å/ps. The force constant used was 200 kcal.mol⁻¹·Å⁻². 505 All the reaction mechanisms were simulated in both forward and reverse directions, and 506 in each case the reported free energy profile corresponds to the optimal combination of 507 both, as usually done when applying this strategy. The reaction coordinate for each 508 profile was always the difference between the donor atom to P and P to acceptor 509 (attacking atom) distances (See Extended Data Table S2 for values)

510 Cloning, Protein Expression and Purification

511 Cloning and protein purification were performed as described previously^{7,9,28}. 512 Briefly, plasmid pQE80_DesKC_{DEST} was generated by subcloning DesKC_{DEST} from 513 pHPKS/Pxyl-desKDEST⁸ into pQE80-DesR through PCR amplification using primers 514 DesK_BamHI_F (CACGGATCCAGCAAGGAGCGCGAACGACTTG) and DesK_Sall_R 515 (TCCTGGTCGACTTA TTTTGAATTATTAGGAATTGC), BamHI and Sall digestion, and 516 ligation. pQE32-DesKC_{0193A} constructed using primers DesKQ193A was 517 (GATACGCTTGGGGGCAAAGCTTT CTC) and DesK_Rev 518 (GAATTATTAGGAATTGCCATGGTAAGCTTGGTC) by RF cloning⁵⁵. pQE80 DesR-519 REC_{Q10A} was generated by similar procedure using DesR Q10A F (5'-520 GTATATTTATTGCAGAAGATGCGCAAATGCTGCTGG-3') DesR-REC R (5'-521 GCTTCGCTGTATAAGTCCTCCATCAG-3'). Recombinant proteins DesKC, DesR-REC 522 and derived mutants were expressed as N-terminally His6-tagged fusions in E. coli 523 strain TOP10F'. The last purification step was a size exclusion chromatography (HiLoad 524 16/60 Superdex 75 preparation grade column; GE) equilibrated with 20 mM Tris-HCl pH 525 8.0, 0.3 M NaCI (SEC buffer). All proteins were concentrated to ~10 mg/mL and stored 526 at -80°C.

527 Autodephosphorylation, Phosphotransferase and Phosphatase Assays

528 Phosphorylation of DesKC was performed by incubating purified DesKC with 10 mM 529 ATP and 10 mM MgCl₂ for an hour at 24°C in SEC buffer. The reaction mix was then 530 loaded in a Superdex75 10/300 column (GE Healthcare) equilibrated in with a solution 531 containing the SEC buffer. To obtain phosphorylated DesR_{REC} and DesR_{REC-Q10A}, each 532 pure protein (~600 µM) was autophosphorylated using 50 mM acetyl phosphate and 30 533 mM MgCl₂ for an hour at 24°C, in the same buffer. Reactions were stopped by adding 534 EDTA (50 mM), and loaded into a Superdex75 10/300 column (GE Healthcare) 535 equilibrated in buffer SEC. The peak corresponding to the dimeric species was selected 536 for further analysis.

The phosphotransfer assay was started by incubating P~DesKC (wild type or mutant) at a concentration of 26 μ M (concentration of the monomeric species) with equimolar amounts of DesR_{REC} (wild type or mutant) in reaction buffer (20 mM Tris-HCl pH 8.0, 0.3 M NaCl and 5 mM MgCl₂) for 30' at 24°C. At different time points the reactions were stopped by adding SDS-PAGE sample buffer with 2.5 mM of DTT. Excess 542 DTT was blocked with 40 mM iodo-acetamide and loaded in a Phos-tag SDS-PAGE, as 543 described before²⁸. Coomassie blue-stained gels were scanned and quantification was 544 done by densitometry using ImageJ⁵⁶.

Auto-dephosphorylation of P~DesR_{REC} (wild type or mutants) was performed as above, incubating P~DesR_{REC}, at different protein concentrations, in Reaction buffer at 24°C. On the other hand, DesK promoted DesR dephosphorylation was done by incubating 26 μ M P~DesR_{REC} in the presence of DesKC (wild type or mutants) at 26 μ M (except for DesKC_{H188V} that different concentrations were tested) in Reaction buffer at 24°C. Reactions were stopped at different time points and loaded in a Phos-tag SDS-PAGE as described above.

552 **Protein Crystallization, Data Collection and Model Building**

553 The DesKC_{H188E}:DesR_{REC-Q10A} complex was prepared by mixing 300 μ M DesKC_{H188F} and 554 165 µM DesR_{BEC-010A} in a buffer containing Tris-HCl 20 mM pH 8.0, 0.3M NaCl, 20 mM 555 MgCl₂ and 5 mM AMP-PCP (non-hydrolysable analogue of ATP). The complex 556 crystallized in a mother liquor containing 20% (w/v) PEG 3350 and 0.35 M tri-potassium 557 citrate⁷. Protein drops were setup by mixing 2 µL of protein plus 2 µL of mother liquor. 558 Cryo-protection was achieved by quick soaking in 20% (w/v) PEG 3350, 0.35 M tri-559 potassium citrate, 5 mM AMP-PCP, 25% (v/v) glycerol, and 20 to 150 mM MgCl2 + 5 560 mM BeF3⁻.

561 Single crystal X-ray diffraction was performed in a copper rotating anode home source 562 (Protein Crystallography Facility, Institut Pasteur de Montevideo). Diffraction data was 563 processed with autoProc⁵⁷ and structure was solved by molecular replacement using 564 each molecule of 5IUK⁷ as search probe in Phaser⁵⁸. We reprocessed the X-ray 565 diffraction data from 5IUN by merging two dataset from the same crystal in autoPROC. 566 Model building was done in Coot⁵⁹ and refinement in Buster⁶⁰. Validation was done 567 throughout and towards the end of refinement using MolProbity tools⁶¹. Visualization of 568 protein models and structural analyses and figure rendering were performed with 569 Pymol⁶². Software for data processing, structure determination and analysis was 570 provided by the SBGrid Consortium³⁰.

571 Analytical Size Exclusion Chromatography

To analyze phosphorylation-triggered dimerization of Q10A_{RR} substitution, recombinant purified proteins dimeric phosphorylated $DesR_{REC}$ and $DesR_{REC-Q10A}$ Dimeric species were obtained as described above and degree of phosphorylation for both proteins was determined by Phostag SDS-PAGE. Then 100µL of 100 µM protein, with a degree of phosphorylation of 60%, was loaded in a Superdex75 10/300 column (GE Healthcare) equilibrated with a buffer containing 20 mM Tris-HCl pH 8.0, 0.3 M NaCl and 10 mM MgCl₂ and run at 0.5 mL/min.

579 Kinetic Model of the DesK-DesR System

580 Several kinetic models were constructed taking into consideration biochemical, 581 structural and biophysical information from previous work^{7,9,28}. We considered models 582 with two or three functional states of DesK, in the latter case assuming the kinase state 583 to be different from the phosphotransferase state. The autophosphorylation reaction 584 was not included in the model, since all the reactions were performed in the absence of 585 ATP or ADP. In addition, we tested models in which the binding of DesK might occur 586 with identical or different affinity when DesR is phosphorylated or not. We also tested a 587 simplified model where we discarded the second binding site for DesR in the 588 phosphotransferase state of DesK. The best-fitted model is depicted in Fig. 5. Each 589 model was constructed as custom scripts in Matlab R2020a (MathWorks Inc.), 590 consisting in a set of differential equations that describes each reaction, conformational 591 rearrangement (like kinase/phosphotransferase transition) or interaction between 592 proteins. The model equations were solved numerically using the ode15s solver. 593 Parameters were globally optimized by minimizing the sum of squared residuals (SSR) 594 of the full dataset with the simplex search method as implemented in fminsearch with 595 boundaries (John D'Errico (2021). fminsearchbnd, 596 fminsearchcon (https://www.mathworks.com/matlabcentral/fileexchange/8277-

597 fminsearchbnd-fminsearchcon), MATLAB Central File Exchange). Given the complexity 598 of the parameter space, an exhaustive search was performed, starting from 800000 599 different initial conditions, uniformly distributed in the multidimensional space. The 600 confidence intervals of the fitted parameters were estimated using lsqcurvefit in Matlab. 601 In order to compare different kinetic models the standard deviation of the SSR was 602 calculated by boostrap. Models that showed a SSR worse than 10 standard deviations 603 with respect to the best model were discarded.

604 Isothermal Titration Calorimetry

Isothermal titration calorimetry (ITC) assay was performed on a VP-ITC (Microcal VP-ITC (Malvern Panalytical) as previously described⁷. Briefly, titration was carried out at 15°C in a buffer containing 20 mM Tris-HCl pH 8.0, 0.3 M NaCl, 10 mM MgCl2 and 0.5 mM AMP-PNP. The concentration of dimeric DesKC was 15 μ M and DesR_{REC} 380 μ M. Raw data were analyzed with NITPIC v1.2.7^{63,64} and integrated binding isotherms were fitted to a model with two independent sequential sites using SEDPHAT v12.1b⁶⁵. Plots including corrected thermograms, fittings of binding isotherms and residuals were done 612 with GUSSI⁶⁶.

613 Data availability

The X ray structures presented have been deposited in the wwPDB with accession codes 7SSJ (DesK-DesR complex in the phosphatase state) and 7SSI (DesK-DesR_{Q10A} complex in the phosphotransfer state). Raw X ray diffraction data corresponding to each one of these structures are publicly available at SBGrid Data Bank (http://data.sbgrid.org) as dataset entries XXX.

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624 Competing interests

625 The authors declare no competing interests.

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796		

797 Table 1: Free energy derived from QM/MM calculations

Phosphotransfer	∆G [‡] (kcal/mol)	∆Gº (kcal/mol)
DesK:DesR	7.1	1.8
DesK:DesR (no Mg ²⁺)	19.4	8.2
DesK:DesR (T80A)	7.9	1.1
YPD1:SLN1-R1	7.1	4.8
His:Asp (water)	25.1	5.0

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800 Figure 1: HK phosphatase-catalyzed reaction. a) Reaction center of the DesK-DesR 801 complex in the phosphatase state. The gray mesh shows 2mFobs-DFcalc electron 802 density map contoured at 1, and the green mesh shows positive peaks of the mFobs-803 DFcalc electron density contour at 3.5. Key residues are depicted in sticks. b) 804 Bidimentional free energy diagram energy as a function of the bonds broken and 805 formed for the phosphatase reaction. The distance between the phosphorus atom of 806 the phosphoryl-moiety and the OE2 atom of D54_{RR} (P-Asp O) against the distance 807 between the phosphorus and hydroxyl anion (P-OH) is shown in the plot as a way of describing the reaction path. Inset: QM region for QM-MM simulations of the 808 809 phosphatase reaction. c) Free Energy profile (orange) and main interatomic distances as 810 a function of the reaction coordinate for the phosphatase reaction, defined as the 811 difference between the P-Asp O distance and the OH-P distance.

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813 Figure 2: Phosphoryl-transfer reaction. a) Reaction diagram of the phosphoryl-814 transfer reaction. The left panel shows the bidimentional free energy diagram energy as 815 function of the breaking and forming bonds of the DesK-DesR system (AspO-P vs 816 HiseN-P). Right panel shows the Free Energy diagram for the reversible phosphorelay SIn1-Ypd1 systems. **b)** Phosphotransfer kinetics in the absence of Mg²⁺. Equimolar 817 818 amounts of phosphorylated DesR_{REC} and DesKC were incubated and analyzed by 819 densitometry from Coomasie stained Phostag SDS-PAGE. The solid lines shows the 820 exponential fit of the phosphorylation degree of DesR-REC (red) and DesKC (blue).



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Figure 3: Analysis of highly conserved residues in the phosphoryl-transfer reaction. Assays were performed with DesKC~P in the presence of DesR_{REC-T80A} (a) or DesR_{REC} (b); or DesR_{REC}~P in the presence of DesKC_{D189A} (c) or DesKC (d). Degree of phosphorylated of DesKC (red) and DesR_{REC} mutants (blue) was analyzed *in vitro* by densitometry of Coomasie stained PhosTag SDS gels. Two independent experiments are shown for each reaction.



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Figure 4: Modulation of phosphoryl-transfer reversibility. (a) Structural comparison between the phosphotransferase (residues depicted in light blue) and phosphatase (in green) complexes highlighting the interaction established by $R84_{RR}$. (b) Phosphoryltransfer assay showing the distribution of the phosphoryl-moiety between DesKC and phosphorylated DesR_{REC-R84A}. (c) Similar view as in (a), showing Q10_{RR} inserted in a pocket created at the ATP binding domain:DHp interface, and interacting with K194_{HK}. (d) Phosphoryl-transfer assay of DesKC and phosphorylated DesR_{REC-Q10A}.

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Figure 5: Phosphoryl-transfer assay dataset and model fitting. (a) Schematic representation of the kinetic model. Yellow shaded box are all the reactions involved in the HK-catalyzed dephosphorylation reaction (H188V module). The red box includes the dimerization and autodephosphorylation of the RR module. The gray box corresponds to the H188E module, which includes RR binding to the phosphotransferase, without phosphoryl-transfer reaction. Green box describes the phosphoryl-transfer reactions,

843	and in cyan the HK conformational transitions are represented. (b) Parameters of the
844	best fitted model. (c to g) Phosphoryl-transfer assays. The reactions were performed
845	using either DesR_{REC} (c, e and g) or $\text{DesR}_{\text{REC-Q10A}}$ (d and f) and started with
846	phosphorylated wild type DesKC (c and d), $\text{DesKC}_{\text{Q193A}}$ (e and f) or $\text{DesKC}_{\text{DEST}}$ (g). (h)
847	Phosphatase assay incubating phosphorylated $DesR_{REC}$ alone, with $DesKC_{H188V}$ or
848	$\text{DesKC}_{\text{H188E}}$. The continuous trace (red and blue lines) depicts the prediction of the best-
849	fitted model.



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854 Figure 6: DesK-DesR structural and kinetic model (a). b) Diagrammatic representation 855 of the free energy transition of the phosphoryl-transfer reaction of a more loose (left panel) or tight (right panel) TS and the role of the Mg2+. A shorter His-Asp distance 856 857 places the phosphoryl moiety at interaction distance to the coordinated Mg⁺² at the 858 beginning of the reaction, thereby stabilizing the initial state c) Schematic 859 representation of the conformational equilibrium of DesKC in the absence of 860 phosphorylation, indicating the higher energy of the phosphotransferase state. d) 861 Proposed role of R235_{HK} and K242_{HK} in sensing the phosphorylation state of the His.